This article was downloaded by: On: 22 January 2011 Access details: Access Details: Free Access Publisher Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



**To cite this Article** Zhang, Zhen , Gao, Zhu-Lin , Fang, Xin , Wang, Yue-Hu , Xiao, Huai , Hao, Xiao-Jiang , Liu, Guang-Ming and He, Hong-Ping(2008) 'Two new triterpenoid saponins from *Glochidion puberum*', Journal of Asian Natural Products Research, 10: 11, 1029 – 1034

To link to this Article: DOI: 10.1080/10286020802318826

URL: http://dx.doi.org/10.1080/10286020802318826

# PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



## Two new triterpenoid saponins from Glochidion puberum

Zhen Zhang<sup>ab</sup>, Zhu-Lin Gao<sup>b</sup>, Xin Fang<sup>b</sup>, Yue-Hu Wang<sup>b</sup>, Huai Xiao<sup>a</sup>, Xiao-Jiang Hao<sup>b</sup>, Guang-Ming Liu<sup>a</sup> and Hong-Ping He<sup>b</sup>\*

<sup>a</sup>College of Pharmacy, Dali University, Yunnan, China; <sup>b</sup>State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunning Institute of Botany, Chinese Academy of Science, Yunnan, China

(Received 12 December 2007; final version received 12 June 2008)

Two new triterpenoid saponins, puberosides A and B, were isolated from the aerial parts of *Glochidion puberum*. Their structures were elucidated as  $3\alpha$ -[(O- $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  3)-O- $\alpha$ -L-arabinopyranosyl)oxy]-22 $\alpha$ -benzoyloxy-olean-12-ene-16 $\alpha$ ,28-diol (1) and  $3\alpha$ -[(O- $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  3)-O- $\alpha$ -L-arabinopyranosyl)oxy]-28-benzoyloxy-olean-12-ene-16 $\alpha$ ,22 $\alpha$ -diol (2) by analysis of their spectroscopic data.

Keywords: Glochidion puberum; triterpenoid saponins; puberoside A; puberoside B

#### 1. Introduction

Glochidion puberum (Linn.) Hutch., belonging to the family Euphorbiaceae, grows in tropical and subtropical forests. In China, this plant has been used for the treatment of feverish illness, malaria, dysentery, gastroenteritis, and so on [1]. Investigation of *Glochidion puberum* has been rarely carried out. So far, only 12 compounds, such as friedelin, lup-20 (29)-ene-1 $\beta$ -ol-3 $\alpha$ -yl acetate, have been isolated [2].

The biological and therapeutic significance of this plant stimulated us to intensively study *Glochidion puberum* and two new triterpenoid saponins were isolated. This paper reports the isolation and the structural elucidation of the new compounds **1** and **2**.

#### 2. Results and discussion

Compound 1 was obtained as an amorphous powder with a molecular formula of  $C_{48}H_{72}O_{14}$  analyzed from its HRESIMS. The IR spectra showed absorption bands for hydroxyl group

 $(3417 \text{ cm}^{-1})$ , carbonyl group  $(1702 \text{ cm}^{-1})$ , and aromatic ring (1603, 1585, and 1451 cm<sup>-1</sup>) [3,4]. The  $^{13}$ C NMR spectrum of **1** showed 48 signals, of which 30 were assigned to the triterpene aglycone moiety. The chemical shifts of two olefinic carbons at  $\delta$  124.2 and 143.5 and comparison of the remaining chemical shifts (Table 1) with those for pentacyclic triterpenes  $(22\alpha$ -hydroxyerythrodiols) revealed the presence of the characteristic signals of an olean-12-ene triterpene [5]. The methylene resonance at  $\delta$  64.7 was diagnostic for the hydroxylated C-28, which corresponded to two proton signals at  $\delta$  3.67 and 3.99 in the HSQC spectrum. The absence of a methylene resonance at high field (ca.  $\delta$  15.0) [5,6] in the <sup>13</sup>C NMR spectrum and the appearance of the signal at  $\delta$  4.30 in the <sup>1</sup>H NMR spectrum of **1**, which correlated with C-28 in the HMBC spectrum, suggested the existence of 16-OH in 1 (Figure 2). The correlations between H-16 and H-28 in the ROESY spectrum revealed H-16 to be  $\beta$ -configuration, which indicated  $\alpha$ -configuration of 16-OH. The lack of another

<sup>\*</sup>Corresponding author. Email: hehongping@mail.kib.ac.cn

$\begin{array}{c c c c c c c c c c c c c c c c c c c $		1			2			030
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		С	Н	HMBC $(H \rightarrow C)$	С	Н	HMBC $(H \rightarrow C)$	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		38.3 (t)			39.8 (t)	1.63 (m)		
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		27.0 (t)	1.88 (m)		27.1 (t)			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		90.4 (d)		23, 24, 1"	90.4 (d)			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$				· ·				
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$			0.80 (brs)			0.80 (brs)		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			1.65 (m)					
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		33.9 (t)	1.45 (m)		33.6 (t)	1.40 (m)		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		41.0 (s)			41.2 (s)			Ņ
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$			1.64 (m)			1.58 (m)		Ŋ
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1	37.8 (s)			37.9 (s)			lar
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		24.7 (t)	1.99 (brs)		24.7 (t)			Zhang et al.
$\begin{array}{cccccccccccccccccccccccccccccccccccc$						5.32 (brs)		et
$\begin{array}{cccccccccccccccccccccccccccccccccccc$					143.0 (s)			al.
$\begin{array}{cccccccccccccccccccccccccccccccccccc$								
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		37.7 (t)			37.6 (t)			
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		69.4 (d)		28	69.4 (d)			
$\begin{array}{cccccccccccccccccccccccccccccccccccc$								
$\begin{array}{cccccccccccccccccccccccccccccccccccc$			2.40 (dd, $J = 13.6, 4$ )			2.63 (brs)		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		. ,	1.94 (m)			1.88 (m)		
21 39.9 (t) 1.94 (m) 38.3 (t) 1.77 (2H, m)   1.00 (m) 100 (m) 100 (m) 100 (m)   22 72.1 (d) 5.91 (brs) 16, 18, 20, 7' 72.1 (d) 5.90 (brs) 18   23 28.6 (q) 1.07 (3H, s) 3, 5 27.9 (q) 1.07 (3H, s) 3   24 17.0 (q) 0.86 (3H, s) 3, 5 17.0 (q) 0.86 (3H, s) 3		31.0(s)	1.2 ( ( ))		31.0(s)	1.20 (III)		
22 72.1 (d) 5.91 (brs) 16, 18, 20, 7' 72.1 (d) 5.90 (brs) 18   23 28.6 (q) 1.07 (3H, s) 3, 5 27.9 (q) 1.07 (3H, s) 3   24 17.0 (q) 0.86 (3H, s) 3, 5 17.0 (q) 0.86 (3H, s) 3						1.77 (2H, m)		
23 28.6 (q) 1.07 (3H, s) 3, 5 27.9 (q) 1.07 (3H, s) 3   24 17.0 (q) 0.86 (3H, s) 3, 5 17.0 (q) 0.86 (3H, s) 3		72.1 (d)		16, 18, 20, 7'	72.1 (d)	5.90 (brs)	18	
24 17.0 (q) 0.86 (3H, s) 3, 5 17.0 (q) 0.86 (3H, s) 3								
		· 1/					3	
25 16.2 (q) 0.99 (3H, s) 9, 15 16.2 (q) 0.98 (3H, s)		16.2 (q)	0.99 (3H, s)	9, 15	16.2 (q)	0.98 (3H, s)	5	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$								

1030

Table 1.  ${}^{1}$ H (400 Hz),  ${}^{13}$ C (100 Hz), and HMBC NMR spectral data of compounds 1 and 2 (CH<sub>3</sub>OH, ppm).

Table 1 – continued

		1		2		
	С	Н	HMBC $(H \rightarrow C)$	С	Н	HMBC $(H \rightarrow C)$
27	27.9 (q)	1.29 (3H, s)	8, 13, 15	27.9 (q)	1.30 (3H, s)	13
28	64.7 (t)	3.67 (brs) 3.99 (d, <i>J</i> = 11.2)	22	64.7 (t)	4.69 (1H, d, <i>J</i> = 11.2) 4.33 (1H, d, <i>J</i> = 11.2)	18 22, 7′
29	34.3 (q)	0.93 (3H, s)	19, 21	34.3 (q)	0.92 (3H, s)	
30 1'	27.6 (q) 132.1 (s)	1.03 (3H, s)	19, 21	27.5 (q) 132.0 (s)	1.01 (3H, s)	
2', 6'	130.5 (d)	8.04 (d, $J = 7.3$ )	4', 7'	130.6 (d)	8.02 (d, $J = 7.3$ )	4', 7'
3', 5'	129.7 (d)	7.49 (dd, $J = 7.3, 7.6$ )		129.5 (d)	7.45 (dd, $J = 7.3, 7.4$ )	
4'	134.1(d)	7.60 (t, $J = 7.3, 7.6$ )		134.1 (d)	7.57 (t, $J = 7.3, 7.4$ )	
7′	167.2 (s)			168.4 (s)		
1″	107.1 (d)	4.28 (d, $J = 7.6$ )	3	107.1	4.27 (d, $J = 7.6$ )	3
2″	72.1 (d)	3.74 (brs)		72.1 (d)	3.70 (m)	
3″	83.9 (d)	3.67 (brs)		83.8 (d)	3.63 (m)	
4″	69.5 (d)	4.05 (brs)		69.5 (d)	4.00 (brs)	
5″	66.7 (t)	3.89 (brs)		66.7 (t)	3.84 (m)	
		3.59 (brs)			3.54 (m)	
1‴	105.4 (d)	4.55 (d, $J = 7.2$ )	3″	105.4 (d)	4.55 (d, $J = 7.2$ )	3″
2‴	75.3 (d)	3.32 (brs)		75.3 (d)	3.27 (m)	
3‴	77.6 (d)	3.42 (brs)		77.6 (d)	3.28 (brs)	
4‴	71.1 (d)	3.37 (brs)		71.1 (d)	3.31 (m)	
5‴	77.9 (d)	3.33 (brs)		77.9 (d)	3.27 (m)	
6′′′	62.3 (t)	3.81 (m)		62.3 (t)	3.81 (m)	
		3.71 (m)			3.67 (m)	

Z. Zhang et al.

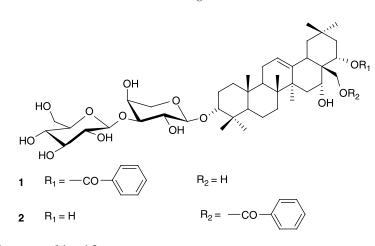


Figure 1. Structures of 1 and 2.

methylene resonance at ca.  $\delta$  33.0 (C-22) [6] and the appearance of HMBC correlations between the proton at  $\delta$  5.91 (brs) and C-16, C-18, C-20 (Figure 2) indicated that a hydroxyl group should be allocated to C-22. The correlations between H-16 and H-22 in the ROESY spectrum revealed H-22 to be  $\beta$ -configuration, which indicated  $\alpha$ -configuration of 22-OH. Moreover, the <sup>1</sup>H and <sup>13</sup>C NMR, DEPT, and HMBC spectra showed the existence of a benzoyl group at  $\delta_{\rm H}$  8.04 (d, 2H, J = 7.3 Hz, H-2', 6'), 7.49 (dd, 2H, J = 7.3and 7.6 Hz, H-3', 5') and 7.60 (t, J = 7.3 and 7.6 Hz, H-4'), and at  $\delta_c$  132.1 (s), 130.5 (d), 129.7 (d), 134.1(d), 129.7 (d), and 130.5 (d) (Table 1 and Figure 1) [7], which was linked to C-22 by the HMBC correlation between C-7' and H-22 (Figure 2).

The carbon at  $\delta$  90.4 in the <sup>13</sup>C NMR spectrum, showing a significant glycosidation shift, suggested the linkage of the sugar

moiety to C-3 of the aglycone [8]. H-3  $(\delta 3.13)$  correlated with H-24  $(\delta 0.86)$  in the ROESY spectrum, indicating the  $\alpha$ -configuration of C-3-OH. By means of HSQC, HMBC, and <sup>1</sup>H-<sup>1</sup>H COSY spectra, compound 1 had two monosaccharides, arabinopyranose ( $\delta_{\rm C}$  107.1, 72.1, 84.2, 69.5, and 66.7), and glucopyranose ( $\delta_{\rm C}$  105.4, 75.3, 77.6, 71.1, 77.9, and 62.3) [9,10], respectively. The anomeric protons appearing at  $\delta$ 4.28 (1H, d, J = 7.6 Hz) and  $\delta$  4.55 (1 H, d, J = 7.2 Hz), and their corresponding carbons resonating at  $\delta_{\rm C}$  107.1 (C-1") and 105.4 (C-1<sup>///</sup>) from HSQC experiment, suggested the presence of  $\alpha$ -L-arabinopyranosyl and  $\beta$ -D-glucopyranosyl groups [9,10]. The HMBC correlation between H-1" [ $\delta$  4.28 (1H, d, J = 7.6 Hz)] and C-3 ( $\delta_{\text{C}}$  90.4) confirmed that the arabinopyranosyl unit was linked at C-3. The HMBC correlation between H-1<sup>*III*</sup> [ $\delta$  4.55 (1 H, d, J = 7.2 Hz)]

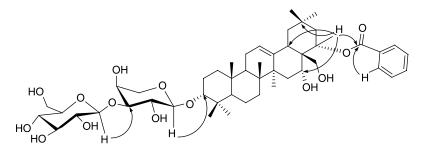


Figure 2. Key HMBC correlations of **1**.

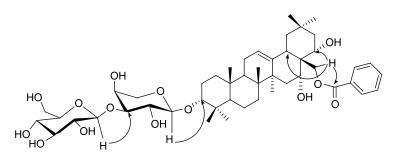


Figure 3. Key HMBC correlations of 2.

and C-3" ( $\delta_{\rm C}$  83.9) revealed the 1  $\rightarrow$  3 linkage of the two sugars. Thus, the structure of **1** was established as  $3\alpha$ -[(O- $\beta$ -D-glucopyranosyl-(1  $\rightarrow$  3)-O- $\alpha$ -L-arabinopyranosyl)oxy]-22 $\alpha$ benzoyloxy-olean-12-ene-16 $\alpha$ ,28-diol.

Compound 2 was also obtained as an amorphous powder with a molecular formula of C<sub>48</sub>H<sub>72</sub>O<sub>14</sub> analyzed from its HRESIMS. The IR spectrum showed absorption band for hydroxyl group  $(3416 \text{ cm}^{-1})$ , carbonyl group  $(1715 \text{ cm}^{-1})$ , and aromatic ring (1603, 1587, and  $1452 \text{ cm}^{-1}$ ) [2,3]. Comparison of the <sup>1</sup>H and  ${}^{13}C$  spectral data of 2 with those of 1 indicated, similar to 1, that compound 2 also included two monosaccharides and a benzoyl-substituted triterpenoid moiety. The difference between 1 and 2 was the benzoyl moiety that was allocated at C-28 ( $\delta_{\rm C}$  64.7) in 2, not C-22, and this can be confirmed by the HMBC correlations between H-28 and C-7' (Figure 3). So, the structure of 2 was established as  $3\alpha$ -[(O- $\beta$ -D-glucopyranosyl- $(1 \rightarrow 3)$ -*O*- $\alpha$ -L-arabinopyranosyl)oxy]-28benzoyloxy-olean-12-ene-16 $\alpha$ ,22 $\alpha$ -diol.

#### 3. Experimental

#### 3.1 General experimental procedures

Optical rotations were carried out by using a Horiba SEAP-300 spectropolarimeter. UV spectra were measured on a Shimadzu 2401 PC. IR spectra were measured as KBr pellets by using a 577 spectrometer. FABMS and HRESIMS data were recorded by using a VG Auto Spec-3000 spectrometer and an API QSTAR PULSAR, respectively. 1D NMR spectra were recorded by using a Bruker AM-400 and 2D NMR spectra were recorded by using a Bruker DRX-500 instrument in which TMS was used as an internal standard for all measurements. Semipreparative HPLC was performed by using an Agilent 1100 liquid chromatograph equipped with a Zorbax SB-C<sub>18</sub>, 9.4 mm  $\times$  25 cm column. Column chromatography was performed by using silica gel (200-300 mesh; Qingdao Marine Chemical Ltd. Co., Qinggao, China) and Sephadex LH-20 (Amershan Biosciences, Switzerland). Fractions were monitored by TLC (Qingdao Marine Chemical Ltd Co., Qingdao, China), and spots were visualized by heating silica gel plates sprayed with 10% H<sub>2</sub>SO<sub>4</sub> in EtOH. Solvents were distilled before use.

#### 3.2 Plant material

The aerial parts of *Glochidion puberum* were collected in Xishuangbanna, Yunnan province of China, in March 2004. The plant was identified by Professor De-Ding Tao (Kunming Institute of Botany, Chinese Academy of Sciences), and a voucher specimen (Kun NO. 0615419) is deposited at the State Key Laboratory of Phytochemistry and Plant Resources in West China, Kunming Institute of Botany, Chinese Academy of Sciences.

### 3.3 Extraction and isolation

The dried aerial parts of *G. puberum* (6 kg) were extracted with hot 95% EtOH. The EtOH extract was successively partitioned between  $H_2O$  and petroleum ether and between  $H_2O$  and EtOAc. The residue from the EtOAc layer (50 g) was subjected to silica

Z. Zhang et al.

gel column chromatography and eluted with gradient petroleum ether–acetone to give four fractions (1–4). Fraction 1 was submitted to Sephadex LH-20 and eluted with MeOH to yield two fractions A (200 mg) and B (100 mg). Fraction A was separated by semipreparative HPLC (CH<sub>3</sub>CN-H<sub>2</sub>O, 60:40), yielding **1** (25 mg,  $t_{\rm R}$  12 min) and **2** (10 mg,  $t_{\rm R}$  8 min).

## 3.3.1 $3\alpha$ -[(O- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-O- $\alpha$ -L-arabinopyranosyl)oxy]-22 $\alpha$ -benzoyloxyolean-12-ene-16 $\alpha$ , 28-diol (1)

Amorphous powder;  $[\alpha]_D^{23.6} + 19.67$  (*c* 0.5, CH<sub>3</sub>OH); UV (CH<sub>3</sub>OH)  $\lambda_{max}$  (log  $\varepsilon$ ) 349 (2.20), 280 (3.56), 273 (3.64), 229 (4.78), 202 (4.80), and 194 (4.48) nm; IR (KBr)  $\nu_{max}$  3417, 1702, 1603, 1585, and 1451 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data see Table 1; FABMS *m/z* 871 [M - H]<sup>-</sup>; HRESIMS *m/z* 871.4857 [M - H]<sup>-</sup> (calcd for C<sub>48</sub>H<sub>71</sub>O<sub>14</sub>, 871.4843).

## 3.3.2 $3\alpha$ -[(O- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-O- $\alpha$ -L-arabinopyranosyl)oxy]-28-benzoyloxyolean-12-ene-16 $\alpha$ , 22 $\alpha$ -diol (2)

Amorphous powder;  $[α]_D^{23.6} - 19.67$  (*c* 0.5, CH<sub>3</sub>OH); UV (CH<sub>3</sub>OH)  $λ_{max}$  (log ε) 272

(3.70), 227 (4.73), and 202 (4.79) nm; IR (KBr)  $\nu_{\text{max}}$  3416, 1715, 1603, 1587, and 1452 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR spectral data see Table 1; FABMS *m/z* 871 [M – H]<sup>-</sup>; HRESIMS *m/z* 871.4828 [M – H]<sup>-</sup> (calcd for C<sub>48</sub>H<sub>71</sub>O<sub>14</sub>, 871.4843).

#### References

- Quan Guo, Zhong Cao, and Yao Ming Jian, *People's Medical Publishing House*, 380, 2000.
- [2] W.H. Hui and M.M. Li, *Phytochemistry* 17, 156 (1978).
- [3] G.V. Subbaraju, D. Rajasekhar, J. Kavitha, and J.I. Jimenez, *Indian J. Chem.* 40, 313 (2001).
- [4] S.H. Day, N.Y. Chiu, L.T. Tsao, J.P. Wang, and Y.C. Lin, J. Nat. Prod. 63, 1560 (2000).
- [5] K. Tori, S. Seo, A. Shiaoka, and Y. Tomita, *Tetrahedron Lett.*, 4227 (1974).
- [6] S.B. Mahato and A.P. kundu, *Phytochemistry* 37, 1517 (1994).
- [7] K.N. Scott, J. Am. Chem. Soci., 8564 (1972).
- [8] Z. Jiang, J.F. Gallard, M.T. Adeline, V. Dumontet, M.V. Tri, T. Sevenet, and M. Païs, *J. Nat. Prod.* 62, 873 (1999).
- [9] R. Srivastava and D.K. Kulshreshtha, *Phy-tochemistry* 25, 2672 (1986).
- [10] N. Li, A.Q. Jia, and Y.Q. Li, Acta Bot. Yunnan 25, 241 (2003).